



Designation: D 7168 – 05

Standard Test Method for ⁹⁹Tc in Water by Solid Phase Extraction Disk¹

This standard is issued under the fixed designation D 7168; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method describes a solid phase extraction (SPE) procedure to separate ⁹⁹Tc from environmental water samples (that is, non-process and effluent waters). Technetium-99 beta activity is measured by liquid scintillation spectrometry.

1.2 This test method is designed to measure ⁹⁹Tc in the range of approximately 0.037 Bq/L (1.0 pCi/L) or greater for a one litre sample.

1.3 The values stated in SI units are to be regarded as standard. Values given in parentheses are provided for information purposes only.

1.4 This method has been used successfully with tap water. It is the user's responsibility to ensure the validity of this test method for samples larger than 1 L and for waters of untested matrices.

1.5 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For specific hazard statements, see Section 9.

2. Referenced Documents

2.1 ASTM Standards:²

D 1129 Terminology Relating to Water

D 1193 Specification for Reagent Water

D 2777 Practice for Determination of Precision and Bias of Applicable Methods of Committee D19 on Water

D 3370 Practices for Sampling Water from Closed Conduits

D 4448 Guide for Sampling Ground-Water Monitoring Wells

D 5847 Practice for Writing Quality Control Specifications for Standard Test Methods for Water Analysis

D 6001 Guide for Direct-Push Water Sampling for Geoenvironmental Investigations

¹ This test method is under the jurisdiction of ASTM Committee D19 on Water and is the direct responsibility of Subcommittee D19.04 on Methods of Radiochemical Analysis.

Current edition approved June 1, 2005. Published June 2005.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

3. Terminology

3.1 *Definitions*—For definitions of terms used in this test method refer to Terminology D 1129.

4. Summary of Test Method

4.1 A measured aliquant of sample is transferred to a beaker. Hydrogen peroxide is added to facilitate the formation of the extractable pertechnetate ion. The sample may be heated to oxidize organics if such are suspected to be present. The entire sample is passed through a technetium-selective SPE disk onto which the pertechnetate is adsorbed. The disk is transferred to a liquid scintillation vial, cocktail added, and the contents well mixed. The beta emission rate of the sample is determined by liquid scintillation spectrometry. Chemical yield corrections are determined by the method of standard additions.

4.2 Minor differences in processing between Extraction Chromatographic Resin Discs and PTFE Membrane Disks are addressed in Variations A and B of the method.

5. Significance and Use

5.1 This method has not been evaluated for all possible matrices. Method suitability should be determined on specific waters of interest.

6. Interferences

6.1 Suspended materials must be removed by filtration or centrifuging prior to processing the sample. Suspended particulate matter in the sample will be physically trapped in part or in whole on or in the SPE extraction material. This may lead to potential inclusion of radionuclide bearing solids or to signal quenching in the liquid scintillation measurement.

6.2 Technetium-99 activity in the sample may overwhelm the signal from the ⁹⁹Tc spike addition and interfere with accurate determination of chemical yield. Samples for which the unspiked sample count rate exceeds 50 % of the spiked sample count rate should be reprepared with an appropriately adjusted aliquant and spike addition levels to minimize contributions to uncertainty in the determination of the chemical yield.

6.3 Organic compounds present in significant quantities in the sample may degrade the extraction performance of the SPE disk or may lead to elevated levels of quench during liquid scintillation analysis. After the addition of hydrogen peroxide, the sample may be heated to destroy trace organic matter in the

sample. If organic components are present in the sample which may survive the peroxide digestion, these may be removed with an appropriate organic removal resin or disk (such as Amberchrom resin or disk) prior to passing the sample through the extraction chromatographic resin disc.

6.4 The disk may retain tritium-labeled compounds. Setting the ^{99}Tc counting window above the maximum energy for the tritium beta particle will eliminate potential tritium interference.

6.5 Elevated levels of nitrates ($> 10\,000$ ppm) will interfere with uptake of ^{99}Tc .

6.6 The higher energy region above the maximum energy for ^{99}Tc should be monitored to help identify cases of significant actinide interference.

6.7 Elevated levels of radionuclides present in anionic form such as iodate, iron (III) and antimony may interfere with measurement of technetium and lead to a positive bias in sample results. Significantly elevated levels of actinides (esp. ^{234}Th decay progeny of uranium) when present in the sample may cause a high bias in the reported ^{99}Tc activity. Manufacturer specific recommendations about interferences should be taken into consideration when determining the applicability of this method for a given matrix.

7. Apparatus

7.1 *Filtering Apparatus*, 47-mm diameter filter apparatus as recommended by the SPE manufacturer.

7.2 *Liquid Scintillation Counter*, with multiple energy region of interest (ROI) capabilities.

7.3 *Scintillation Vials*, 20-mL vials, low potassium glass or plastic, exhibiting suitable optical reproducibility so as not to cause erratic results between samples.

8. Reagents and Materials

8.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to specifications of the Committee on Analytical Reagents of the American Chemical Society.³ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determinations.

8.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean reagent water conforming to Specification **D 1193**, Type III.

8.3 *Radioactive Purity*—Radioactive purity shall be such that the measured radioactivity of blank samples does not exceed the calculated probable error of the measurement.

³ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Annual Standards for Laboratory and National Formulary*, U.S. Pharmaceutical Convention, Inc. (USPC), Rockville, MD.

8.4 *Technetium-Specific Solid Phase Extraction (SPE) Disks or Membranes* (Extraction Chromatographic Resin Discs⁴ or PTFE Membrane Disks^{4,5}).

8.5 *Hydrochloric Acid, 0.5M*—Add 42 mL concentrated HCl to 400 mL of reagent water. Dilute to 1 L with water.

8.6 *Nitric Acid*, concentrated.

8.7 *Hydrogen Peroxide*, 30 %.

8.8 *Technetium-99* as pertechnetate in water or dilute base solution, traceable to a national standards body (such as NIST in the U.S.).

8.9 *Liquid Scintillation Cocktail*—Commercially prepared LSC cocktail or equivalent.^{4,6}

9. Hazards

9.1 Use extreme caution when handling all acids. They are extremely corrosive, and skin contact could result in severe burns.

9.2 When diluting concentrated acids, always use safety glasses and protective clothing, and add the acid to the water.

10. Sampling

10.1 Collect a sample in accordance with Practice **D 3370**, **D 4448** or **D 6001**.

11. Preservation

11.1 Preservation of samples being analyzed for ^{99}Tc is not required.

11.1.1 Samples may be preserved by freezing. Allow samples to come to ambient temperature prior to processing.

11.1.2 Samples may be processed if they have been previously preserved to pH less than 2 with nitric or hydrochloric acid. It is noted that high concentrations of nitric acid will adversely affect chemical yield. Although yield corrections will correct for losses, better results may be obtained by using unpreserved samples.

12. Calibration

12.1 The fractional detector efficiency (ϵ_{Tc}) is determined as outlined in subsequent steps.

12.1.1 Prepare triplicate calibration samples adding at least 20 Bq (~ 540 pCi) of traceable ^{99}Tc in the pertechnetate form to each of three 100 mL portions of reagent water. Each of the three samples is processed using either method variation (A or B), as appropriate.

⁴ The sole source of supply of the Eichrom TEVA® Discs known to the committee at this time is Eichrom Technologies, Inc., Darien, IL. If you are aware of alternative suppliers, please provide this information to ASTM International Headquarters. Your comments will receive careful consideration at a meeting of the responsible technical committee,¹ which you may attend.

⁵ 3M Empore® Tc Rad Disks (3M, St. Paul, MN) have been found satisfactory for this purpose.

⁶ Ultima Gold® LLT (Perkin Elmer Life and Analytical Sciences, 710 Bridgeport Avenue, Shelton, CT 06484-4794) has been found satisfactory for this purpose.