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ANSI/AWWA B503-17
(Revision of ANSI/AWWA B503-11)

AWWA Standard

Sodium Tripolyphosphate

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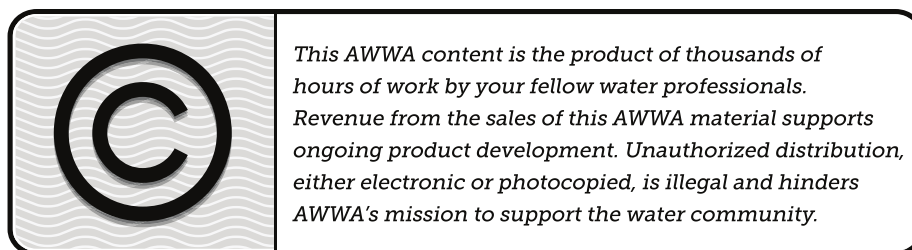
AWWA Standard

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Foreword

This foreword is for information only and is not a part of ANSI /AWWA B503.*

I. Introduction.

I.A. *Background.* Sodium tripolyphosphate ($\text{Na}_5\text{P}_3\text{O}_{10}$) is a manufactured product obtained by combining a comparatively pure chemical—soda ash (Na_2CO_3) or caustic soda (NaOH)—with phosphoric acid (H_3PO_4). The resultant product is subjected to controlled conditions of heating to form a crystalline solid, which is subsequently ground and sized.

Sodium tripolyphosphate is a white solid material, commercially available in granular or powder forms. The density of crushed or granular is approximately 68.7 lb/ft³ (approximately 1,100 kg/m³). A solution of 1 lb/gal (0.12 kg/L) has a specific gravity of approximately 1.10. Refer to safety data sheets (SDSs) available from the supplier or manufacturer for additional information.

Water utilities use sodium tripolyphosphate to control scale and corrosion and to treat red water.

I.B. *History.* The first edition of ANSI/AWWA B503 was developed by the AWWA Committee on Scale and Corrosion-Control Chemicals. It was first published in 1978. Subsequent revisions were approved by the AWWA Board of Directors in 1984, 1989, 1994, 2001, 2005, and 2011. This edition was approved on June 11, 2017.

I.C. *Acceptance.* In May 1985, the US Environmental Protection Agency (USEPA) entered into a cooperative agreement with a consortium led by NSF International (NSF) to develop voluntary third-party consensus standards and a certification program for direct and indirect drinking water additives. Other members of the original consortium included the Water Research Foundation (formerly AwwaRF) and the Conference of State Health and Environmental Managers (COSHEM). The American Water Works Association (AWWA) and the Association of State Drinking Water Administrators (ASDWA) joined later.

In the United States, authority to regulate products for use in, or in contact with, drinking water rests with individual states.[†] Local agencies may choose to impose requirements more stringent than those required by the state. To evaluate the health effects of products and drinking water additives from such products, state and local

* American National Standards Institute, 25 West 43rd Street, Fourth Floor, New York, NY 10036.

† Persons outside the United States should contact the appropriate authority having jurisdiction.

agencies may use various references, including two standards developed under the direction of NSF*: NSF/ANSI 60, Drinking Water Treatment Chemicals—Health Effects, and NSF/ANSI 61, Drinking Water System Components—Health Effects.

Various certification organizations may be involved in certifying products in accordance with NSF/ANSI 60. Individual states or local agencies have authority to accept or accredit certification organizations within their jurisdictions. Accreditation of certification organizations may vary from jurisdiction to jurisdiction.

Annex A, “Toxicology Review and Evaluation Procedures,” to NSF/ANSI 60 does not stipulate a maximum allowable level (MAL) of a contaminant for substances not regulated by a USEPA final maximum contaminant level (MCL). The MALs of an unspecified list of “unregulated contaminants” are based on toxicity testing guidelines (noncarcinogens) and risk characterization methodology (carcinogens). Use of Annex A procedures may not always be identical, depending on the certifier.

ANSI/AWWA B503 addresses additives requirements in Sec. 4.3 of the standard. The transfer of contaminants from chemicals to processed water or to residual solids is becoming a problem of great concern. The language in Sec. 4.3.3 is a recommendation only for direct additives used in the treatment of potable water to be certified by an accredited certification organization in accordance with NSF/ANSI 60, Drinking Water Treatment Chemicals—Health Effects. However, users of the standard may opt to make this certification a requirement for the product. Users of this standard should consult the appropriate state or local agency having jurisdiction in order to

1. Determine additives requirements, including applicable standards.
2. Determine the status of certifications by parties offering to certify products for contact with, or treatment of, drinking water.
3. Determine current information on product certification.

II. Special Issues.

II.A. *Storage and Handling Precautions.* Sodium tripolyphosphate is slightly hygroscopic and must be stored under dry conditions. Exposure to humidity produces a caked condition and accelerates reversion to orthophosphate. This results in poor flow and caking, which interfere with the rate of dilution and effectiveness of the product. For additional information, refer to the MSDS available from the manufacturer or supplier.

* NSF International, 789 North Dixboro Road, Ann Arbor, MI 48105.